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Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information:

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To cite this Article Guo, Yue-Wei and Trivellone, Enrico(2000) 'New Hurghamids from a Red Sea Sponge of the Genus *Hippospongia*', Journal of Asian Natural Products Research, 2: 4, 251 – 256

To link to this Article: DOI: 10.1080/10286020008041363

URL: <http://dx.doi.org/10.1080/10286020008041363>

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NEW HURGHAMIDS FROM A RED SEA SPONGE OF THE GENUS *HIPPOSPONGIA*

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(Received 1 November 1999; Revised 2 November 1999; In final form 23 November 1999)

Three new N-acyl-2-methylene- β -alanine methyl esters, Hurghamides E–G (5–7), were isolated from a Red Sea sponge *Hippospongia* sp. Their structures were elucidated by extensive spectroscopic studies.

Keywords: Sponge; N-acyl-2-methylene- β -alanine methyl esters; Hurghamides E, F and G

INTRODUCTION

We previously reported the isolation and structure elucidation of four N-acyl-2-methylene- β -alanine methyl esters, Hurghamids A–D (1–4) (Fig. 1), from a Red Sea sponge *Hippospongia* sp. [1]. We have now separated three more new analogs, named hurghamides E–G (5–7) (Fig. 2). The isolation and structure elucidation of these new compounds are reported here.

RESULTS AND DISCUSSION

The sponge (500 g, wet weight) was collected by SCUBA off Hurghada in the Red Sea and stored frozen at -80°C until processed. The sponge was

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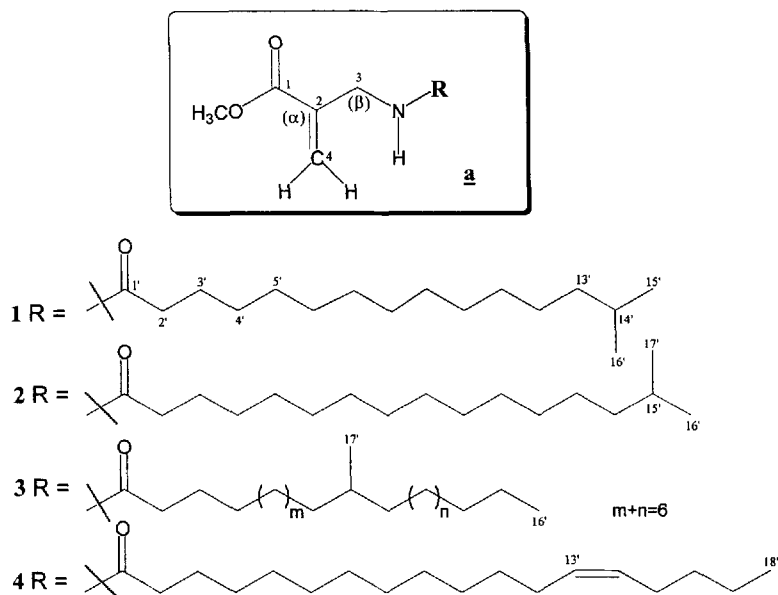


FIGURE 1

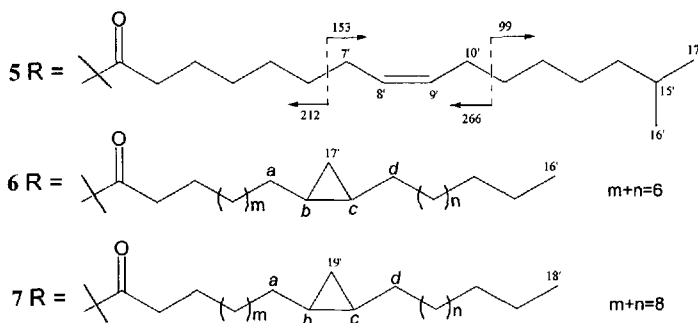


FIGURE 2

exhaustively extracted with Me_2CO , and the extract was partitioned between Et_2O and water. The organic phase after removing the solvent *in vacuo* was subjected to column chromatography on silica gel eluting with petroleum ether- Et_2O system. The major metabolite obtained by this procedure [1] was further separated by reverse-phase HPLC eluting with $\text{CH}_3\text{CN}-\text{H}_2\text{O}$ (70:30) to give hurghamides E-G (5-7) together with known hurghamides reported previously [1].

TABLE I ^{13}C -NMR data^{a, c} of hurghamides E-G (5-7)

Carbon	5	6	7
1	166.5 s	166.5 s	166.5 s
2	136.4 s	136.4 s	136.4 s
3	40.5 t	40.3 t	40.5 t
4	127.4 t	127.1 t	127.4 t
OCH ₃	172.9 s	172.9 s	172.9 s
1'	36.8 t	36.8 t	36.8 t
2'	25.7 t	25.7 t	25.7 t
3'	25.7 t	25.7 t	25.7 t
a	27.2 t	30.2 t	30.2 t
b	129.8 d	15.7 d	15.7 d
c	129.9 d	15.9 d	15.9 d
d	27.1 t	30.2 t	30.2 t
14'	39.0 t	31.9 t	
15'	28.0 d	22.5 t	
16'	22.6 q	13.0 q	31.8 t
17'	22.6 q	10.9 t	22.5 t
18'			13.7 q
19'			10.8 t

^a Bruker AM-500; δ values are reported in ppm referenced to CHCl₃ (δ 77.0); assignments were deduced from the analysis of mono- and hetero-nuclear spectra.

^b The multiplicity was determined by DEPT technique.

^c The methylenes not reported contributed to a large signal centered at δ 29.8.

All of these compounds showed very similar spectroscopic properties. Their IR spectra indicated the presence of a secondary amide (3400 sh, 3420, 1665, 1520 and 1260 cm^{-1}), an ester (1720 cm^{-1}), and a terminal methylene (950 cm^{-1}). Careful analysis of their ^1H - ^1H COSY, HMQC, TOCSY and HMBC led to the characteristic of a common N-acyl-2-methylene- β -alanine methyl ester moiety (partial structure **a**) which was confirmed by the comparison of the ^1H - and ^{13}C -NMR data (Table I) with those of co-occurring hurghamides [1]. In fact, the difference among them happened only in the nature of the acyl part (R).

Hurghamide-E (**5**) had a molecular formula $\text{C}_{22}\text{H}_{39}\text{NO}_3$ as deduced from EIMS spectrum (m/z 365, M^+) and ^{13}C -NMR data. The ^1H -NMR spectrum of **5** is very similar to hurghamide-B (**2**) showing also the presence of a terminal isopropyl group (δ 0.87, d, $J=7$ Hz, 6H) in the molecule. In fact, **5** differs from **1** only by the presence of an isolated double bond (δ 5.35, 2H) in its acyl part. The *cis*-stereochemistry of this double bond was deduced from the ^{13}C -NMR chemical shift of the allylic methylene carbons (δ 27.2, 27.1) [2] while the position of this double bond was tentatively assigned at Δ^8 on the basis of EIMS data. In fact, the EIMS spectrum of **5** displayed, apart from the intense fragments reported previously [1], diagnostic fragments at m/z 153 and 99 indicating the position of the double bond.

Hurghamide-F (**6**), with also the molecule formula $C_{22}H_{39}NO_3$, is an isomer of **5**. The acyl part of **6** had to be unbranched as only a 3H triplet at δ 0.88 was present in the methyl region of the 1H -NMR spectrum. In addition, the alkyl chain (R) contained a cyclopropane ring, instead of a double bond, as shown by the three upfield resonances at δ -0.34 (1H, ddd, $J=5, 5, 4$ Hz, H-17'a), 0.56 (ddd, $J=8, 8, 4$ Hz, H-17'b), and δ 0.65 (2H, m, H-b and H-c). Detailed analysis of 1H - 1H COSY, HMQC, TOCSY and HMBC spectra permitted assignment of the carbons of cyclopropane ring and the two flanking methylene groups as reported in Table I. The large difference between the 1H -NMR chemical shifts of the ring geminal methylene protons is a clear indication of the *cis*-stereochemistry of the ring substituents. Comparison of NMR data with those reported for model compound (plakoside A) [3] confirmed this assignment. To locate the cyclopropane ring chemical degradation was necessary. Unfortunately, scarcity of the material prevented from determination the position of the cyclopropane ring.

Hurghamide-G (**7**) ($C_{24}H_{43}NO_3$, two CH_2 more than that of **6**) showed 1H -NMR spectrum almost identical to that of **6** containing also a *cis*-substituted cyclopropane ring (Table I) and a terminal methyl (H₃-18', δ 0.87). Clearly, **7** is a superior homologue of **6**. The position of cyclopropane ring remains to be determined.

This is the first report of 2-methylene- β -alanine methyl esters, N-acylated by unbranched aliphatic fatty acids containing cyclopropane rings, isolated from a marine source. Interestingly, hurghamides (**1**-**7**) show structural similarity with a sleep-inducing lipid, *cis*-octadecenamide (oleamide) [4]. Further study should be conducted to test if hurghamides possess also sleep-inducing activity and to understand what is the biological role of these metabolites during the life cycle of the animal, as well as their biogenetic origin.

EXPERIMENTAL SECTION

General Experimental Procedures

The IR spectra were recorded on a Bio-Rad FTS 7 spectrometer. 1H - and ^{13}C -NMR spectra were recorded on a Bruker AM-500 (500 MHz for 1H and 125 MHz for ^{13}C) spectrometer. Chemical shifts are reported in ppm referenced to $CHCl_3$ as internal standard (δ 7.26 for proton and δ 77.0 for carbon). 1H - and ^{13}C -NMR assignments were supported by 1H - 1H COSY, HMQC and HMBC experiments. EIMS spectra were recorded on a CARLO

ERBA VG TRIO 2000 instrument. Reverse-phase HPLC purification was performed on a Waters liquid chromatograph using a Uvidec-100-III detector. Both analytical [5 μ , 4.6 mm (I.D.) \times 25 cm] and semi-preparative [5 μ , 10 mm (I.D.) \times 25 cm] columns were SPHERISORB-S5 ODS2. Commercial Merck Si gel plates were used for TLC. The chromatograms were sprayed with 0.1% Ce(SO₄)₂ in 2 N H₂SO₄ and heated at 80°C for 5 min to detect the spots.

Collection of the Biological Material

The specimens were collected by Dr. E. Mollo off Hurghada on the Red Sea using SCUBA-diving at a depth of -10 m. They were immediately frozen and transferred to Naples and kept at -80°C until extraction. Voucher specimens are available for inspection at ICMIB of Naples and at Istituto di Zoologia of Genova.

Extraction and Isolation

A frozen sponge (20 g, dry weight after extraction) was extracted with Me₂CO (2.5 l). After concentration, the aqueous residues were extracted with Et₂O (3 \times 150 ml). The combined ether extracts were taken to dryness, yielding oily residues (1.6 g) which was chromatographed on silica gel column using eluants of increasing polarity from light petroleum ether to Et₂O. The fraction eluting with petroleum ether-Et₂O (3:7) yielded crude hurghamides that was again chromatographed on a silica gel column eluting with the same eluants mentioned above. This procedure yielded formally pure hurghamides (0.26 g, 1.3% dry weight) as a white amorphous wax. Prep. HPLC isolation of Hurghamides E-G (**5**-**7**) was carried out by isocratic elution with CH₃CN-H₂O (70:30). This yielded, in order of increasing polarity, **6** (1.6 mg), **5** (0.9 mg), **7** (1.0 mg) together with other known hurghamides [1]. The flow rate for the separation was 2 ml/min.

Spectral Data of 5 A white amorphous wax. IR (liquid film) ν_{\max} 3260, 2880, 2810, 1715, 1640, 1550, 1460, 1380, 940, 720 cm⁻¹; UV (MeOH) λ_{\max} 205 nm (ϵ 14700); ¹H-NMR δ 4.08 (2H, br d, J = 6 Hz, H₂-3), 6.25, 5.83 (each 1H, br s, H₂-4), 5.93 (1H, m, N-H), 5.34 (2H, m, olefinic protons), 3.78 (3H, s, OCH₃), 2.17 (2H, t, J = 7, 7 Hz, H₂-2'), 2.02 (4H, m, methylenes adjacent to the double bond), 1.62 (2H, m, H₂-3'), 1.16 (2H, m, H₂-13'), 1.52 (1H, m, H-14'), 0.87 (6H, d, J = 7 Hz, H₃-16' and H₃-17'); ¹³C-NMR: see Table I; EIMS m/z (%): 365 (M⁺, 12%), 334 (14%), 306 (15%), 170 (80%), 153 (32%), 142 (42%), 116 (90%), 114 (100%), 99 (25%).

Spectral Data of 6 A white amorphous wax. IR and UV are same as those of **5**; $^1\text{H-NMR}$ δ 4.09 (2H, br d, $J=6$ Hz, $\text{H}_2\text{-3}$), 6.25, 5.83 (each 1H, br s, $\text{H}_2\text{-4}$), 5.92 (1H, m, N-H), 3.78 (3H, s, OCH_3), 2.17 (2H, t, $J=7$, 7 Hz, $\text{H}_2\text{-2}'$), 1.62 (2H, m, $\text{H}_2\text{-3}'$), 1.37, 1.12 (each 2H, m, $\text{H}_2\text{-a}$ and $\text{H}_2\text{-d}$), 0.89 (3H, t, $J=7$, 7 Hz, $\text{H}_3\text{-16}'$), 0.65 (2H, m, H-b and H-c), 0.56 (1H, ddd, $J=8$, 8, 4 Hz, H-17'b), -0.34 (1H, ddd, $J=5$, 5, 4 Hz, H-17'a); $^{13}\text{C-NMR}$: see Table I; EIMS m/z (%): 365 (M^+ , 10%), 334 (14%), 170 (50%), 142 (32%), 116 (90%), 114 (100%).

Spectral Data of 7 A white amorphous wax. IR and UV are same as those of **5**; $^1\text{H-NMR}$ δ 4.09 (2H, br d, $J=7$ Hz, $\text{H}_2\text{-3}$), 6.25, 5.83 (each 1H, br s, $\text{H}_2\text{-4}$), 5.92 (1H, m, N-H), 3.78 (3H, s, OCH_3), 2.17 (2H, t, $J=7$, 7 Hz, $\text{H}_2\text{-2}'$), 1.63 (2H, m, $\text{H}_2\text{-3}'$), 1.37, 1.12 (each 2H, m, $\text{H}_2\text{-a}$ and $\text{H}_2\text{-d}$), 0.89 (3H, t, $J=7$, 7 Hz, $\text{H}_3\text{-18}'$), 0.65 (2H, m, H-b and H-c), 0.56 (1H, ddd, $J=8$, 8, 4 Hz, H-19'b), -0.34 (1H, ddd, $J=5$, 5, 4 Hz, H-19'a); $^{13}\text{C-NMR}$: see Table I; EIMS m/z (%): 393 (M^+ , 13%), 362 (14%), 170 (50%), 142 (37%), 116 (92%), 114 (100%).

Acknowledgements

Dr. Y.W. Guo thanks Dr. G. Cimino and Dr. M. Gavagnin for helpful discussions on the work. Thanks are also due to Dr. E. Mollo, Mr. F. Castelluccio, Mr. G. Scognamiglio and Mr. R. Turco for their valuable technical help. The NMR and the mass spectra were obtained from the "ICMIB-NMR Service" and from the "Servizio di Spettrometria di Massa del CNR e dell'Università di Napoli". The staff of both services is gratefully acknowledged. This research was supported by the Italian-Egyptian bilateral project (No. 13) "Marine Chemistry of Red Sea Suez Canal Algae and Invertebrates" and by the C.N.R. strategic project "Tecnologie Chimiche Innovative".

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